

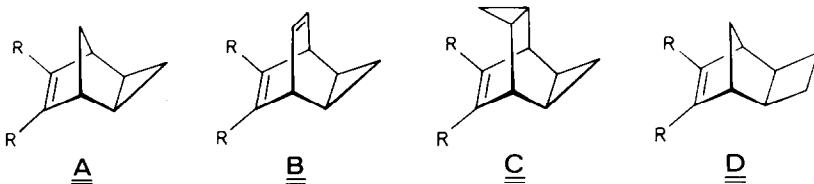
STEREOELECTRONIC ASPECTS OF THE  $[\pi^{2+}\sigma^2]$ -PHOTOCYCLOADDITION REACTION\*\*  
 $[\pi^{2+}\sigma^2]$ -PHOTOCYCLOADDITIONS IN THE TETRACYCLO[4.4.0.0<sup>2,10</sup>.0<sup>3,7</sup>]DEC-4-ENE SYSTEM

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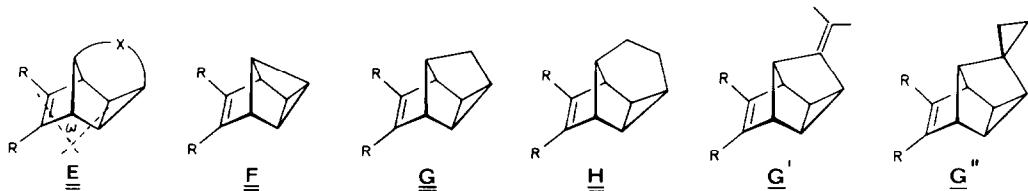
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In agreement with expectations efficient ethylene/cyclopropane cycloadditions have been observed in derivatives of tetracyclo[4.4.0.0<sup>2,10</sup>.0<sup>3,7</sup>]dec-4-ene upon direct excitation.

In line with the intrinsic properties of the participating bonds the ethylene/cyclopropane (oxirane, aziridine) photocycloaddition ( $[\pi^{2+}\sigma^2]$ ) is strongly geometry-dependant: It is efficient in the exo-tricyclooctene framework A, not however in the endo-isomer <sup>1,2)</sup>; it has no chance against the  $[\pi^{2+}\pi]$ -reaction in the homobarrelene B <sup>3)</sup>, occurs uniformly with the exo-cyclopropane ring in the bishomobarrelene C <sup>4)</sup> and it generally proceeds only intramole-



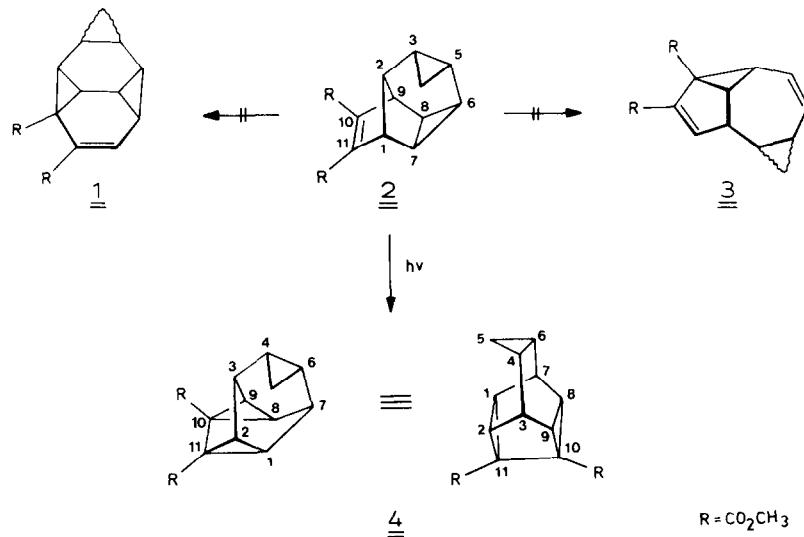
cularly <sup>4,5)</sup>. A comparable ethylene/cyclobutane cycloaddition did not take place even under favourable conditions (D) <sup>5)</sup>. For the series E <sup>5)</sup>, 3,8-bridged analogues of A, the expectations were such, that, independant of the mechanism, the increase in the interplanar angle  $\omega$  from F to H and thereby in the transannular  $\pi/\sigma$ -interaction <sup>6,7)</sup> would favour the transannular bond formations. And indeed, such transformations were realised in G <sup>5)</sup> (not G', G'' <sup>8)</sup>) but not in F <sup>7)</sup>.



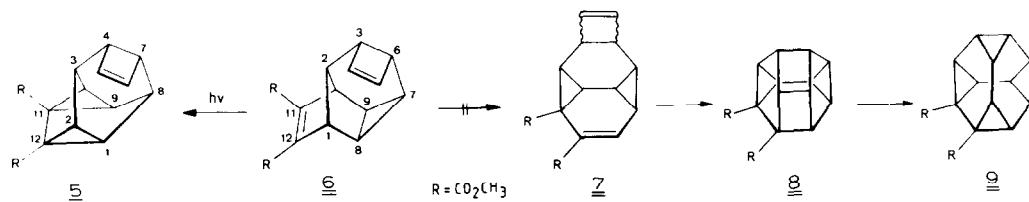
With the synthesis of the pentacyclic diesters 2/6 we now have models for the next higher homologue H (tetracyclo[4.4.0.0<sup>2,10</sup>.0<sup>3,7</sup>]dec-4-ene; the diester ( $R=CO_2CH_3$ ) is not available from dihydrobarrelene and dimethyl acetylenedicarboxylate (ADM) <sup>9)</sup>) whereby the ester groups guarantee favourable excitation <sup>10)</sup>. 2 (dimethyl pentacyclo[5.4.0.0<sup>2,9</sup>.0<sup>3,5</sup>.0<sup>6,8</sup>]undec-10-ene-10,11-dicarboxylate) was obtained in 80-85% yield by [2+2+2]-addition of ADM to homobarrelene (boiling xylene; colourless oil,  $m/e = 260 (M^+)$ ,  $\lambda_{max} (CH_3CN) = 235$  nm ( $\epsilon = 5050$ )). ([2+2+2]-adducts were also obtained with dicyanoacetylene and with phenyltriazolinedione, not, how-

ever, with maleic anhydride or 1,2-bistolylsulphonyl ethylene<sup>11)</sup>). Under the same and varied conditions the Nenitzescu hydrocarbon affords only a moderate yield (15%) of 6 (dimethyl pentacyclo[6.4.0.0<sup>2,10</sup>.0<sup>3,6</sup>.0<sup>7,9</sup>]dodeca-4,11-diene-11,12-dicarboxylate), the cyclobutene unit probably causing side reactions<sup>12)</sup>. 6 had to be separated from polymeric material by chromatography (colourless oil, m/e = 272 ( $M^+$ ),  $\lambda_{max}$  ( $CH_3CN$ ) = 235 nm ( $\epsilon$  = 5040)). With regard to the scope of homo-Diels-Alder-additions<sup>9)</sup> it is worthy of note that bicyclo[2.2.2]octa-5,7-diene 2-one<sup>13)</sup> did not react with ADM, dicyanoacetylene or N-phenyltriazolinedione (up to 200°C).

Upon direct excitation (10<sup>-3</sup> to 10<sup>-4</sup> M, N<sub>2</sub>-sat.  $CH_3CN$ -solutions, 0°C, quartz-filter, high pressure Hg-lamp (150 W)) 2 is quickly and uniformly (TLC, <sup>1</sup>H-NMR) transformed into the cry-



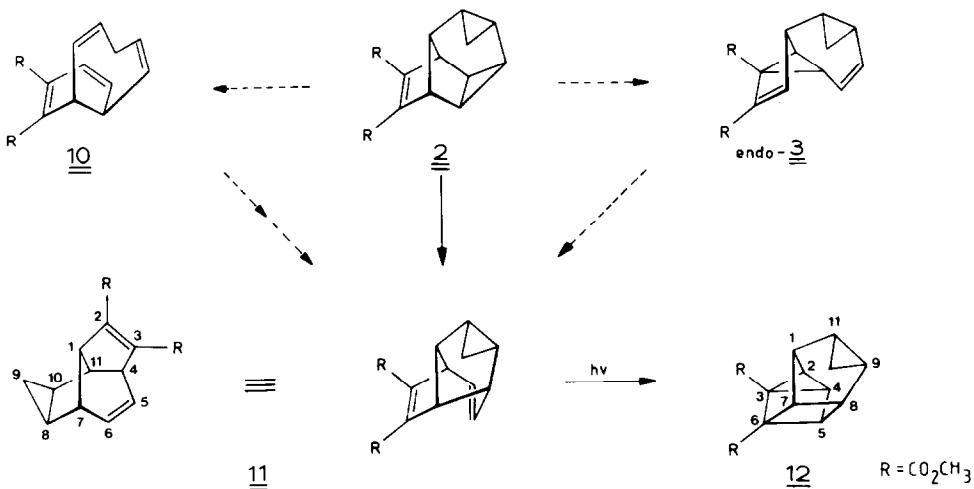
stalline, C<sub>s</sub>-symmetrical [ $\pi^2+\sigma_2$ ]-adduct 4 (dimethyl hexacyclo[5.4.0.0<sup>2,11</sup>.0<sup>3,9</sup>.0<sup>4,6</sup>.0<sup>8,10</sup>]undecane-10,11-dicarboxylate, m.p. 72°C, m/e = 260 ( $M^+$ )). Under these conditions 6 is isomerised with comparable selectivity (>90% isolated) into crystalline 5 (dimethyl hexacyclo[6.4.0.



<sup>0<sup>2,12</sup>.0<sup>3,10</sup>.0<sup>4,7</sup>.0<sup>9,11</sup>]dodec-5-ene-11,12-dicarboxylate, m.p. 81°C, m/e = 272 ( $M^+$ )). The hydrocarbons 4/5 ( $R=H$ ) had been synthesised by other routes<sup>14)</sup>. Qualitatively, the [ $\pi^2+\sigma_2$ ]-additions in 2/6 are faster and more selective than in the lower homologues 5. As in the latter sensitised excitation (acetone) affords only polymeric material. In view of the behaviour of</sup>

G, G' and G'' it is remarkable, that the cyclopropane ring in 2 does not induce concurrent formation of 1 and/or 3. endo-2, the (thermally certainly very labile) product of a 1,3-shift in 2, would be an attractive precursor of the bridged tris- $\sigma$ -homobenzenes 8(9)<sup>15)</sup>.

Being interested in the scope of the [ $\sigma^{2+}$ ] $\sigma^{2+}$ ]-cycloreversion reaction<sup>16)</sup> we also studied the thermal behaviour of 2, which has the structural elements of a cis-tris-[3.1.1]- $\sigma$ -homobenzene (cf. the fast retro-Diels-Alder cleavage in lumibullvalene<sup>17)</sup>). 2 is stable in solution up to at least 200°C. Thermolysis in the vapour phase (450–500°C) leads, however, to a selective transformation: The product, isolated in up to 90% yield (based on extent of conversion) is 11 (dimethyl tetracyclo[5.4.0.0<sup>4,11</sup>.0<sup>8,10</sup>]undeca-2,5-diene-2,3-dicarboxylate, oil, m/e = 260 ( $M^+$ ),  $\lambda_{\text{max}}(\text{CH}_3\text{CN}) = 236 \text{ nm } (\epsilon = 4930)$ ). The structure determined by NMR-analysis (<sup>1</sup>H-, <sup>13</sup>C-) is supported by the photoisomerisation (direct excitation) to 12 (dimethyl hexacyclo[5.4.0.0<sup>2,4</sup>.0<sup>3,6</sup>.0<sup>5,8</sup>.0<sup>9,11</sup>]undecane-3,6-dicarboxylate, 80–88%, m/e = 260 ( $M^+$ )). endo-3



(followed by Cope Rearrangement) rather than 10 is considered to be an intermediate in the surprisingly selective isomerisation 2 → 11.

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\*\*) Photochemical Transformations, part 62; part 61: H. Prinzbach, G. Fischer, G. Rihs, G. Seidelmeier, E. Heilbronner and Yang Z.-z., Tetrahedron Lett. 1982, 1251.

- 1) H. Prinzbach, G.v. Veh and W. Eberbach, Angew. Chem., Internat. Ed. Engl. 4, 436 (1965); P.K. Freeman, G. Kuper and V.N.M. Rao, Tetrahedron Lett. 1965, 3301.
- 2) Rev.: D. Döpp in Houben-Weyl 4/5a, Thieme 1975, p. 443; T. Toda, K. Nakano, A. Yamae and T. Mukai, Tetrahedron 31, 1597 (1975); W. Eberbach and H. Prinzbach, Organic Photochemical Synthesis, Vol. 2, 104 (1976); A. de Meijere, Ch. Weitemeyer and O. Schallner, Chem. Ber.

- 110, 1504 (1977); H. Prinzbach, G. Sedelmeier and H.-D. Martin, Angew. Chem., Internat. Ed. Engl. 16, 103 (1977); G. Sedelmeier, H. Prinzbach and H.-D. Martin, Chimia 33, 329 (1979); R. Srinivasan, J. A. Ors, K. H. Brown, T. Baum, L. S. White and A. R. Rossi, J. Am. Chem. Soc. 102, 5297 (1980).
- 3) H. Prinzbach, H. Fritz, H. Hagemann, D. Hunkler, S. Kagabu and G. Philippoussian, Chem. Ber. 107, 1971 (1974).
- 4) H. Prinzbach, M. Klaus and W. Mayer, Angew. Chem., Internat. Ed. Engl. 8, 883 (1969); H. Prinzbach, S. Kagabu and H. Fritz, Angew. Chem., Internat. Ed. Engl. 13, 482 (1974); S. Kagabu and H. Prinzbach, Tetrahedron Lett. 1975, 29.
- 5) H. Prinzbach and D. Hunkler, Chem. Ber. 106, 1804 (1973); cit. lit; H. Prinzbach, S. Kagabu and H. Fritz, Angew. Chem., Internat. Ed. Engl. 13, 482 (1974); S. Kagabu and H. Prinzbach, Tetrahedron Lett. 1975, 29.
- 6) E.g. P. Bischof, E. Heilbronner, H. Prinzbach and H.-D. Martin, Helv. Chim. Acta 54, 1072 (1971).
- 7) J. Stapersma, Ph. D. Thesis, Free Univ. Amsterdam 1982; cf. A. H. A. Tinnemans and D. C. Neckers, Tetrahedron Lett. 1978, 1713; G. Sugowdz, P. J. Collin and W. H. F. Sasse, Austral. J. Chem. 26, 147 (1973); G. W. Klumpp and J. Stapersma, Tetrahedron Lett. 1977, 747.
- 8) H. Prinzbach, W. Auge and M. Basbudak, Chem. Ber. 106, 1837 (1973).
- 9) G. N. Fickes and T. E. Metz, J. Org. Chem. 43, 4057 (1978); cf. H. E. Zimmermann and G. L. Grunewald, J. Am. Chem. Soc. 86, 1434 (1964).
- 10) H.-P. Schal, Ph. D. Thesis, Univ. Freiburg, 1980. The new compounds are characterised by their spectra (IR,  $^1\text{H}$ -,  $^{13}\text{C}$ -NMR, MS) and elemental analysis.
- 11) O. DeLucchi and G. Modena, Chem. Commun. 1982, 914.
- 12) T. Erden and A. deMeijere, Tetrahedron Lett. 1980, 1837.
- 13) H. Prinzbach, H.-P. Schal and D. Hunkler, Tetrahedron Lett. 1978, 2195; R. Gompper and K.-H. Etzbach, Angew. Chem., Internat. Ed. Engl. 17, 603, (1978).
- 14) T. Sasaki, K. Kanematsu and N. Okamura, J. Org. Chem. 40, 3322 (1975); L. A. Paquette, M. J. Kukla, S. V. Ley and S. G. Traynor, J. Am. Chem. Soc. 99, 4756 (1977).
- 15) L. T. Scott and M. Jones, Jr., Chem. Rev. 72, 181 (1972); J. J. Gajewski, Hydrocarbon Thermal Isomerisations, Academic Press, N. Y. 1981.
- 16) Cf. M. Maas, M. Lutterbeck, D. Hunkler and H. Prinzbach, Tetrahedron Lett. 1983, preceeding paper.
- 17) M. Jones, Jr., S. D. Reich and L. T. Scott, J. Am. Chem. Soc. 92, 3118 (1970).

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